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# 2,2'-Bi(9,9-di-*n*-propylfluorene)

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The central part of the title molecule,  $C_{38}H_{42}$ , is planar, all the rings being in the same plane; the lateral chains (excluding H atoms) are also planar, with each pair almost perpendicular to the ring plane. In the dimer, the two alkyl-substituted fluorene moieties are head-to-foot. The molecule is on a special position of the space group, a centre of symmetry.

### Comment

Oligomers based on fluorene are interesting materials for optoelectronic applications, such as optical limiting, twophoton technology or light-emitting devices (Rheinhart, 1999). Nevertheless, although polyfluorene-based materials exhibit high luminescent yields, and thermal and oxidative stability, in the solid state, their optical properties are strongly influenced by the degree of organization of the material. As a rule, thin layers of regular poly(alkylfluorene) develop an additional broad band in the yellow part of their luminescent



or electroluminescent spectra attributed to the formation of excimers and/or aggregates (Kläner et al., 1999). One way to suppress these troublesome phenomena is to use a well defined crystalline structure of another dimer of fluorene for which conformational motion is avoided. Recently, we have reported the crystal structure of 2,2'-bi(9,9-dihexylfluorene) (Suchod & Stéphan, 2000). As part of our programme, we present here the structure of another dimer of fluorene, (I), for which the alkyl chains grafted at the 9-position of the fluorene moiety have been shortened from 6 to 3 CH<sub>2</sub> units.

Surprisingly, with *n*-propyl chains the molecular conformation is totally different from that with hexyl ones. Whereas

with the latter, alkyl chains were on the same side of the fluorene ring, with propyl chains the fluorene moieties are head-to-foot. The stacking of the molecules is also very different; the planes of conjugated dimers are parallel but shifted, while in the preceding structure there was a perpendicular arrangement of neighbouring molecules. Nevertheless, in both cases, photoluminescence measurements on crystals clearly indicate that the shift leading to yellow instead of blue light emission is avoided.

#### Experimental

The title compound was obtained as described previously (Suchod & Stéphan, 2000) via Ni-catalyzed coupling the corresponding of 2bromofluorene monomeric units.

#### Crystal data

C38H42	Mo $K\alpha$ radiation
$M_r = 498.75$	Cell parameters from 20
Monoclinic, $P2_1/c$	reflections
a = 8.9944 (9)  Å	$ heta=10 ext{-}12^\circ$
b = 16.257 (2)  Å	$\mu = 0.062 \text{ mm}^{-1}$
c = 10.519(1) Å	T = 293  K
$\beta = 103.08 \ (1)^{\circ}$	Parallelepiped, translucent pale
V = 1498.2 (3) Å <sup>3</sup>	white
Z = 2	$0.3 \times 0.3 \times 0.2 \text{ mm}$
$D_x = 1.106 \text{ Mg m}^{-3}$	

## Data collection

wR = 0.045

S = 1.183

3724 reflections

257 parameters

Nonius CAD-4 diffractometer	$h = -12 \rightarrow 12$
$\omega$ scans	$k = 0 \rightarrow 22$
4655 measured reflections	$l = 0 \rightarrow 14$
4355 independent reflections 3724 reflections with $I > 2\sigma(I)$	2 standard reflections every 100 reflections
$R_{\rm int} = 0.01$ $\theta_{\rm max} = 29.96^{\circ}$	intensity decay: 1%
Refinement	
Refinement on F	$w = 1/(0.1 + 1.4\sigma^2)$
R = 0.076	$(\Delta/\sigma)_{\rm max} = 0.027$

 $(\Delta/\sigma)_{\rm max} = 0.027$  $\Delta \rho_{\rm max} = 0.20 \ {\rm e} \ {\rm \AA}^{-3}$  $\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$ Extinction correction: Zachariasen (1967)All H-atom parameters refined Extinction coefficient: 0.85 (3)

Data collection: CAD-4 Software (Enraf-Nonius, 1998); cell refinement: CAD-4 Software; data reduction: Xtal3.2 SORTRF ADDREF (Hall et al., 1992); program(s) used to solve structure: Xtal3.2 GENTAN; program(s) used to refine structure: Xtal3.2 CRYLSQ; molecular graphics: Xtal3.2 ORTEP; software used to prepare material for publication: Xtal3.2 BONDLA CIFIO.

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